

19-METHYL vs. 18-METHYL GROUP MIGRATION IN THE REARRANGEMENT  
OF 3 $\beta$ -ACETOXY-9 $\alpha$ ,11 $\alpha$ -EPOXY-5 $\alpha$ -LANOSTAN-7-ONE

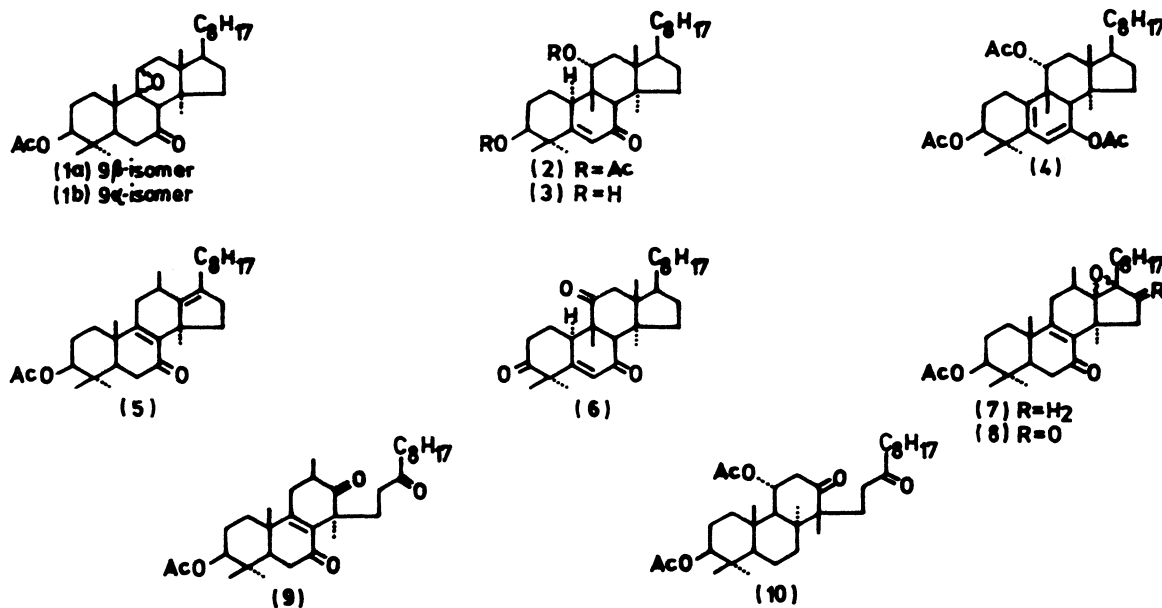
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BF<sub>3</sub>·Et<sub>2</sub>O-catalyzed rearrangement of the keto-epoxide (1b) was found to be a complex reaction leading to compounds (2), (4), and (5) having the rearranged carbon skeletons; spectral and chemical evidences are presented to confirm their structures.

The BF<sub>3</sub>·Et<sub>2</sub>O-catalyzed rearrangement of the epoxide (1a) has been reported<sup>1,2)</sup> to give in high total yield compounds having the cucurbitane skeleton. On the other hand, similar reaction of the 9 $\alpha$ ,11 $\alpha$ -epoxide (1b) afforded compound (5) as the only isolable product.<sup>2)</sup> We found, however, that this rearrangement is more complex reaction and gives products resulting from the two possible modes of the opening of the epoxide ring.

The BF<sub>3</sub>·Et<sub>2</sub>O-catalyzed rearrangement of epoxy-ketone (1b) carried out in acetic anhydride (r.t.; 5 min) gave after isolation and chromatography four products, (2) (mp 148-150°C; yield 13%), (4) (oil; yield 10%), (5)<sup>2)</sup> (yield 44%), and 3 $\beta$ -acetoxy-5 $\alpha$ ,9 $\beta$ -lanostan-7,11-dione<sup>3)</sup> (yield 12%).

The structure of cucurbitane (2) was inferred from its spectral properties [ $\delta$  6.06, 5.18, and 4.85;  $\nu_{\max}$  1750 and 1658 cm<sup>-1</sup>;  $\lambda_{\max}$  244 nm ( $\epsilon$  11600)] and their comparison with the known properties of the 3 $\beta$ ,11 $\beta$ -diacetoxy-19(10 $\rightarrow$ 9 $\beta$ )abeo-10 $\alpha$ -lanost-5-en-7-one.<sup>1)</sup>



The compound (2) was hydrolyzed to (3) (amorphous,  $\delta$  6.06, 3.90, and 3.58;  $\lambda_{\max}$  248 nm) which was oxidized to the triketone (6) [mp 175-178°C;  $\nu_{\max}$  1718, 1698, and 1658  $\text{cm}^{-1}$ ;  $\lambda_{\max}$  248 nm ( $\epsilon$  8800);  $m/z$  454( $M^+$ ), 439, 411, 291, and 205] identical with the product obtained from (2) ( $11\beta$ -OAc isomer)<sup>1)</sup> by similar transformations.

The structure of compound (4) follows from the spectral data;  $m/z$  584( $M^+$ );  $\delta$  5.86(s, 6-H), 5.15(m,  $11\beta$ -H), 4.71(t, J 6.5Hz,  $3\alpha$ -H), 2.09, 2.06, and 2.02 (three  $\text{CH}_3\text{COO}$ ); and  $\lambda_{\max}$  275 nm. Compound (4) was probably formed via  $3\beta$ -acetoxy-19(10 $\rightarrow$ 9 $\beta$ ) abeo-lanost-5(10)-en-7-one, since pure (2) could not be transformed to (4) when treated with  $\text{BF}_3 \cdot \text{Et}_2\text{O}$  in acetic anhydride.

The structure of compound (5) which was proposed solely on the basis of its spectral properties<sup>2)</sup> is now supported by chemical transformations. Epoxidation of (5) with *m*-chloroperbenzoic acid gave epoxide (7) [mp 162-164°C;  $\delta$  4.49, 2.49, and 2.05;  $\nu_{\max}$  1728, 1658, 1610, and 1255  $\text{cm}^{-1}$ ;  $\lambda_{\max}$  251 nm ( $\epsilon$  10000);  $m/z$  498, 385, 372, 164, and 149] while oxidation with  $\text{CrO}_3$  in AcOH afforded low yield of the D-ring epoxy-ketone (8) [mp 172-174°C;  $\lambda_{\max}$  247 nm ( $\epsilon$  12500);  $\nu_{\max}$  1728, 1715, and 1665  $\text{cm}^{-1}$ ;  $m/z$  512, 496, 481, and 301]. The oxidation of the allylic position 11 was not detected in contrast to the easy oxidation of  $3\beta$ -acetoxy- $5\alpha$ -lanost-8-en-7-one to the respective 8-en-7,11-dione.

Ruthenium tetroxide - sodium metaperiodate oxidation of (5) yielded the seco-compound (9) [oil, yield 58% after preparative TLC;  $\delta$  4.53(m,  $3\alpha$ -H), 3.12 (br s,  $11\beta$ -H), 3.02(br s,  $11\alpha$ -H), and 2.07(s,  $\text{CH}_3\text{COO}$ );  $\nu_{\max}$  1728, 1715, 1663, and 1620  $\text{cm}^{-1}$ ;  $\lambda_{\max}$  248 nm ( $\epsilon$  8200)]. Mass spectral fragmentation of compound (9) was similar to those of seco-compound (10)<sup>4)</sup> and contained all major fragments arising from McLafferty rearrangement as well as from  $\alpha$ - and  $\beta$ -cleavages, characteristic of saturated ketones. Additionally, fragment  $m/z$  57, arising from the rupture of  $\text{C}_{13} - \text{C}_{14}$  and  $\text{C}_{11} - \text{C}_{12}$  bonds was very intense (71%).

The presented results indicate that the described rearrangement proceeds via C-9 and C-11 carbocationic species. It is proposed therefore, that electronic and conformational factors<sup>5)</sup> compete in directing the course of the rearrangements of the epoxides (1).

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